Environmental Management of Firefighting Foam

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Firefighting foam management overview

- Life and safety have always been paramount considerations
- Foam types & impacts
- Emerging worldwide concerns for FOCs
- Queensland risk review and Policy
- Risk reduction measures
- Practical transition to Best Practice



Firefighting foam users

- Qld Fire & Emergency Services (Fluorine-free, non-persistent foams since 2003)
- Bulk fuels & chemicals facilities
- Defence facilities (air bases & vessels)
- Airports & transport (often close to wetlands & waterways)
- Maritime (wharves, ships, fire-fighting tugs, stockpiles)
- Mining sector (large vehicles, currently uncontained releases)
- Offshore platforms (over water, uncontained)
- Foam hand extinguishers (waste disposal issues)





Department of Environment and Heritage Protection



FOAM USE IS HIGHLY DISPERSIVE



Fire fighting foam composition

FOAM CONCENTRATE



CONCERNS

- AQUATIC BIOTA
- WATER POLLUTION
- SOIL CONTAMINATION
- HUMAN HEALTH

IMPACT MECHANISMS SHORT-TERM

- O₂ DEPLETION IN WATERWAYS
- ACUTE TOXICITY (low & short-term)

*<u>LONG-TERM</u>

- PERSISTENCE
- BIOACCUMULATION
- CHRONIC TOXICITY (long-term)

FOAM USE IS HIGHLY DISPERSIVE



Firefighting Foams CHARACTERISTICS

FOAM USE IS HIGHLY DISPERSIVE

- Difficult to contain
- High solubility (dissolve & move in water)
- High mobility in soils (to groundwater)

FLUORINATED SURFACTANTS

- Developed in 1960s
- Toxic compounds that do not degrade
- Many bioaccumulate
- Previously regarded as benign
- Worldwide dispersal, highly mobile
- Alternatives available since 2003-2005
- CHRONIC (long-term) EFFECTS





Government

PROBLEMS FROM INTENSIVE USE SITES

- e.g. Oakey 1.2ML AFFF foam released over 25 years. FOAM USE IS HIGHLY DISPERSIVE
- High solubility (dissolve & move in surface & ground water)
- High mobility in soils (to waterways and groundwater)



Short-chain C4-C6 plume likely to be well ahead of C8 PFOS extent

Fluorinated Organic Compounds (FOCs/PFCs)

• <u>Persistence</u> is extreme (non-degradable)

Ρ

B

- Bioaccumulative for many compounds
- <u>Toxicity</u> (mostly chronic long-term effects concerns)
- Environmental & health concerns known since ~2000
- 100s of fluorinated compounds in foams
- No alternatives to PFCs for foam until ~2002-2006
- <u>All</u> fluorinated organics are toxic and problematic
- Many are precursors to PFOS and PFOA or similar



Why now? - Pollutants of emerging concern

(e.g. Asbestos - Ozone-depleting CFCs - DDT pesticide - Leaded petrol - Diesel particulates - etc.)

- Persistent toxic fluorinated organics used in some foams.
- Rapidly emerging indications of long-term adverse effects for human health & wildlife worldwide.
- > Numerous legacy contaminated sites (& ongoing releases).
- Very high cleanup costs (e.g Airport A\$150M, truck fire A\$1.2M).



Figure 2.1 A - Number of published papers on perfluoroalkyl substances 1994 onwards (Trojanowicz & Koc [8] & Buck *et al* [9]) Rapidly growing (public) information on fluorinated compounds in peerreviewed scientific publications since 2008. Grandjean & Clapp 2015

>2,500 papers on PFASs from 2001-2011 Trojanowicz & Koc 2013

Industry knowledge (cancer) since 1997 US (Ohio) PFOA compensation trial evidence in 2015



Why now? - Pollutants of emerging concern

(e.g. Asbestos – Ozone depleting - DDT pesticide - Leaded petrol - Diesel particulates – etc.)

- PFOS and PFOA are only 2 of ~200 compounds of concern.
- Hundreds of FOCs in or resulting from fluorinated foam.
- Plus numerous unknown transformation products.
- New analysis method for detecting hidden compounds.

Parfluorobutana sulfanamida amina# (C4)	6.2 fluorotolomor thiosther amido sulfania agid
Perfluoroputane sulfonamide amine* (C4)	4:2 fluorotelomer unoetner annuo sunonic acid
ernuoropentale sunonannue annue (CS)	4.2 Indototeloiner sunonannde betaine
erfluorohexane sulfonamide amine aarborulie asid# (C4)	5:1:2 fluorotelomer betaine
Perindorobulane suffonantide antino carboxytic acid. (C4)	5.5 Intoroteioner betane
Perfluoropentane sulfonamide amino carboxylic acid* (C5)	7:1:2 fluoroteiomer betaine
Perfluorohexane sulfonamide amino carboxylic acid* (C6)	7:3 fluorotelomer betaine
Perfluorohexane sulfonamide ammonio dicarboxylic acid*	8:2 fluorotelomer sulfonamide amine
Perfluoropentane sulfonamide ammonio dicarboxylic acid*	8:2 fluorotelomer sulfonamide betaine
* Perfluorosulfonamides are precursors for sulfonate compounds)	8:2 fluorotelomer thio hydroxy ammonium
Perfluorohexane sultonic acid (C6 PFHxS)	8:2 fluorotelomer thioether amido sulfonic acid
Perfluoroheptane sulfonic acid (C7 PFHpS)	8:2 fluorotelomer thioether amino carboxylic acid
Perfluorooctane sulfonic acid (C8 PFOS)	9:1:2 fluorotelomer betaine
5:2 fluorotelomer sulfonamide amine	9:3 fluorotelomer betaine
5:2 fluorotelomer sulfonamide betaine	10:2 fluorotelomer sulfonamide betaine
5:2 fluorotelomer thio hydroxy ammonium	10:2 fluorotelomer thioether amino carboxylic acid
5:2 fluorotelomer thioether amido amino carboxylic acid	
Fable 2.1 A – Fluorinated organic compounds (FOCs) in sam Backe, Day & Field (2013)–FOCs in AFFF and groundwater [ples of firefighting foams, groundwater & surfactants 32]
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(Perfluoroheptyl sulfonate (C7 PFHepS)) (Perfluorooctyl sulfonate (C8 PFOS))

Perfluoropentyl carboxylate (C5 PFPeA)

rfluorohexyl carboxylate (C6 PFHxA)

D'Agostino and Mabury (2014)-No	wel FOCs in 10 AFFF concentrates ar	nd 2 commercial fluorinat	ed surfactants. [34]
Groups and classes of PFASs	Dominant form	Number of compo	unds Total
Perfluoroalkylamido betaine (PFAAB)-related (A, B, C)			
n=3, 4, 5, 67, 8, 9, 10, 11, 12, 14	n=8, C14H14ON2F17 +1	(11 compounds)	
n=6, 7, 8, 9, 10, 11, 12, 14	n=8, C16H16O3N2F17 +1	(8 compounds)	
n=8, 10, 12	n=8, C ₁₉ H ₂₂ O ₃ N ₂ F ₁₇ +1	(3 compounds)	(22 compounds)
Fluorotelomerthioalkylamido be	taine (FTSAB)-related (D, E, F, G	, H)	
n=4, 6, 8, 10	n=6, C ₁₀ H ₆ O ₂ SF ₁₃ ⁻¹	(4 compounds)	
n=4, 6, 8, 10, 12	n=6, C15H20ON2SF13 *1	(5 compounds)	
n=4, 6, 8, 10, 12, 14	n=6, C17H22O3N2SF13 *1	(6 compounds)	
n=6, 8, 10	n=6, C16H22ON2SF13 *1	(3 compounds)	
n=6	n=6, C ₁₆ H ₂₂ O ₂ N ₂ SF ₁₃ *1	(6 compounds)	(19 compounds)
Fluorotelomermercaptoalkylamido sulfonate (FTSAS)-related (I, J, K)			
n=4, 6, 8, 10, 12, 14	n=6, C15H17O4NS2F13	(6 compounds)	
n=4, 6, 8, 10	n=6, C15H17O5NS2F13 -1	(4 compounds)	
n=4, 6, 8	n=6, C ₁₁ H ₈ O ₂ SF ₁₃ ⁻¹	(3 compounds)	(13 compounds)
Fluorotelomersulfonamide alkyl	betaine (FTAB)-related (L, M)		
n=6, 8, 10, 12, 14	n=6, C ₁₅ H ₂₀ O ₄ N ₂ SF ₁₃ ⁺¹ [48]	(5 compounds)	
n=6, 8, 10	n=6, C ₁₃ H ₁₈ O ₂ N ₂ SF ₁₃ *1	(3 compounds)	(8 compounds)
Fluorotelomer betaine (FTB) (N	,0)		
n=5, 7, 9, 11, 13	n=5, C ₁₂ H ₁₅ O ₂ NF ₁₁ ⁺¹	(5 compounds)	
n=5, 7, 9, 11, 13, 15	n=5, C ₁₂ H ₁₄ O ₂ NF ₁₂ *1	(6 compounds)	(11 compounds)
Fluorotelomerthiohydroxyl ammonium (FTSHA) (P, Q)			
n=6, 8, 10	n=6, C14H19ONSF13 +1	(3 compounds)	
n=6, 8	n=6, C ₁₄ H ₁₉ O ₂ NSF ₁₃ ⁺¹	(2 compounds)	(6 compounds)
Perfluoroalkylsulfonamido-based surfactants (PFASA-) (R/S, T, U, V) (Likely C3-C9 sulfonate precursors)			
n=3, 4, 5, 6, 7, 8 (branched)	n=6, C ₁₄ H ₁₈ O ₄ N ₂ SF ₁₃ ⁺¹	(6 compounds)	
n=3, 4, 5, 6	n=6, C ₁₄ H ₁₈ O ₄ N ₂ SF ₁₃ ⁺¹	(4 compounds)	
n=3, 4, 5, 6, 7, 8	n=6, C ₁₁ H ₁₄ O ₂ N ₂ SF ₁₃ ⁺¹	(6 compounds)	
n=3, 4, 5, 6 (branched)	n=6, C17H22O6N2SF13 +1	(4 compounds)	
n=6, 7, 8, 9	n=8, C17H22O6N2SF13 +1	(4 compounds)	(24 compounds)
n = perfluorinated carbon chain let	ngths of C3 to C15. (For A to V refe	r to original paper). To	tal 103 compounds
Table 2.1 D			
Herzke, Olsson and Posner (2012)	-PFASs in consumer products in N	orway (results for 2 curren	nt AFFF) [35].
Perfluorobutanoate PFBA (C4) Perfluorocta	ne sulfonamide PFOSA	(PFOS precursor)
Perfluoropentanoate PFPA (C5) 6:2 fluorotel	omer sulfonate 6:2 FtS	
Perfluorohexanoate PFHxA (C6) 6:2 fluorotel	omer alcohol 6:2 FtOI	ł
Perfluorooctanoate PFOA (C8) 8:2 fluorotel	omer alcohol 8:2 FtOI	I (PFOA precursor)
Perfluorodecanoate PFDA (C10) 10:2 fluorote	lomer alcohol 10:2 FtC	H (PFDA precursor)
Dealling and American DED. A (i	210)		



E.g. one compound can result in 16 transformation and 5 end-products.



Tables of the main known FOCs. (Explanatory Notes)

Regulatory position (the Qld Policy) The *Precautionary Principle (in decision making)* Obligation in legislation triggered by:

- ✓ threat of serious <u>or</u> irreversible environmental damage; and
- ✓ scientific uncertainty as to the nature and scope of the threat of environmental damage.

Justice Preston (2006) clarified considerations and stated: "The function of the precautionary principle is, therefore, to **require the decision-maker to assume that there is, or will be, a serious or irreversible threat of environmental damage** and to take this into account, notwithstanding that there is a degree of scientific uncertainty about whether the threat really exists."

> Chief Judge of the NSW Land and Environment Court, Justice Preston in Telstra Corporation Limited v Hornsby Shire Council [2006] NSWLEC



PFCs exposure poses real threats

Elimination in humans (t_{1/2}):

- C8, PFOS 5.4 years
- C8, PFOA 2.3 to 3.8 yrs
- C6, PFHxS 8.5 years (≈C8)
 (! x 5 half lives [↓] 15-40 years)
- ~200 similar compounds.
- Extensive information now published about adverse effects and behaviour.
- Adverse effects of increasing exposure to combinations emerging.



Environmental Management of Foam

- Review of issues & Policy development (2013-16)
- Foam industry has not self-regulated
- Regulatory Strategy (Policy) model to address
- Prompt, staged implementation needs to occur



END-USER RESPONSIBILITY

Queensland Government

Worldwide concerns about <u>all</u> FOCs

Helsingør & Madrid Statements 2014 - (177 scientists signed) Scientific community concerns:

- Health & environmental impacts of FOCs
- Widespread occurrence of fluorinateds
- Extreme persistence
- Lack of decline & increasing exposure
- Impacts of fluorinated alternatives
- Lack of info & testing for 100s of FOCs
- Lack of transparency by manufacturers
- Synergistic effects likely but unknown
- World-wide & tighter regulation needed
- Problematic & costly waste disposal
- Strongly suggest cease use of all FOCs
- Develop non-toxic alternatives



ARTICLE INFO	ABSTRACT
Article history:	In this discussion paper, the transition from long-chain poly- and perfluorinated alkyl substances
Received 18 March 2014	(PFASs) to fluorinated alternatives is addressed. Long-chain PFASs include perfluorsalityl carbonylic
Available online 14 June 2014	acids (PECAs) with 7 or more perfluorinated carbons, perfluoroalicyl sulfonic acids (PECAs) with 6 or more perfluorinated carbons, and their precursors. Because long-chain PEASs have been found
Handling Editor: J. de Boer	to be persistent, bioaccumulative and toxic, they are being replaced by a wide range of fluorinated alternatives. We summarize key concerns about the potential impacts of fluorinated alternatives on
Tenerali:	human health and the environment in order to provide concise information for different
PROA	stakeholders and the public. These concerns include, amongst others, the likelihood of fluorinated
PEOS	alternatives or their transformation products becoming ubiquitously present in the global environ-
PIE chemicals	ment; the need for more information on uses, properties and effects of fluorinated alternatives;
Runnated surfactants	the formation of persistent terminal transformation products including PFCAs and PFSAs; increasing
Rustinated polymers	environmental and human exposure and potential of adverse effects as a consequence of the high
	ultimate persistence and increasing usage of fluorinated alternatives: the high societal costs that
	would be caused if the uses, environmental fate, and adverse effects of fluorinated alternatives
	had to be investigated by publicly funded research: and the lack of consideration of non-persistent
	alternatives to long-chain PFASs.
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	Example ("http://institute.com/instances on Transaction of a VT 0/1

THE MADRID STATEMENT

The Madrid Statement documents the scientific concensus regarding the persistence and potential for harm of poly- and perfluoreality isobstances (PFASs), and lays out a roadmap to gather needed information and prevent further harm.

The Madrid Statement on Poly- and Perfluoroalkyl Substances (PFASs)

As scientists and other professionals from a variety of disciplines, we are concerned about the production and release into the environment of an increasing number of poly- and perfluoroally! substances (PFASs) for the following reasons:

- PFASs are manuade and found everywhere. PFASs are highly persistent, as they contain perfluorinated chains that only degrade very slowly; if at all, under environmental conditions. It is documented that some polyfluorinated chemicals break down to form perfluorinated ones²⁰.
- PFASs are found in the indoor and ourdoor environment, wildlife, and human tissue and body fluids all over the globe. They are emitted via industrial processes and military and freefabilities and environment our of common environment environment of the soundout



Operational Policy content (Other contaminants considered separately)

Non-persistent foams (e.g. fluorine-free)

- BOD & acute toxicity concerns for waterways
- Contain wastes on site if possible
- Treat or biodegrade wastes on site is possible
- Disposal to trade waste or sewer (WWTP) OK
- Biodegradable and largely self remediating
- Emergency direct releases tolerable
- No significant restriction on dispersed rural firefighting or limited essential testing of critical systems (e.g. dockside to waterways)



(C8) PFOS

Operational Policy content (Other contaminants considered separately)

Persistent foams (e.g. all fluorinated, AFFF, FP FFFP...)

- Know what you have in stocks, asses the risk
- PFOS not acceptable, take out of service now
- ≥C7 foams Put in place interim containment measures & phase out (ASAP <u>within</u> 3 yrs)
- Fully contain wastes in impervious bunding
- C6-<u>PURE</u> foam OK (99.5% certified) <u>BUT</u> wastes must be fully contained in impervious bunding
- (C8) PFOA
- Persistent foams not to be used for testing
 - High temperature destruction of wastes



VALUES AT RISK & COSTS

- Human health (long-term)
- Aquatic ecosystems (short and long-term)
- Primary produce contamination & food chain
- Adjacent/downstream users of land and waters
- Economic, land & amenity values (already evident)
- Land contamination very high cleanup costs
- Cost recovery? Industry, insurers, community...?



Challenges for transition to best practice

Balanced decisions based on ALL considerations:

- Life and safety are paramount
- Re-engineering systems
- Interim containment measures
- *Decontamination of fluorinated systems
- Whole-of-life costs (incl. wastes)
- Long-term containment & management
- Land contamination assessment & remediation

*Cross-contamination of replacement foam by fluorinated residues has already emerged as a very significant and costly problem overseas.



Managing foam use

Life & Safety Considerations are PARAMOUNT

Environmental considerations need to be taken seriously in balanced decision making

Forward planning:

Facilities and contingency planning for containment.

Incident management:

Contain and manage firewater & wastewater.

Waste disposal:

Wastewater disposed of as regulated waste.



Managing foam use

Foam end-user activities:

- Legislative provisions have applied since 1995.
- Policy provides guidance on compliance & priorities.
- Transition to non-persistent foam ASAP for direct releases.
- User need to assess risks & compliance.
- Alternative products certified & demonstrated effective. (high-end users – airports, oil platforms, fuel terminals, brigades, IMO, etc.)
- All fluorinated foam including C6 must be fully contained.

Management measures:

- Choice of suitable C6-PURE or non-persistent product.
- Procedures to prevent releases (contingency plans).
- Containment of firewater and wastewater.
- Disposal of firewater as regulated waste.



overnment



Mythbusting – "Foam can be contained by floating booms" (recent US industry guideline). Foams are soluble and dissolve into the water column !

